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electric current. The part to be coated acts as the cathode (negative electrode) of an electrolytic cell; the electrolyte is a solution of a salt of the metal to be coated; and the

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anode (positive electrode) is usually either a block ...

Electroplating - Wikipedia

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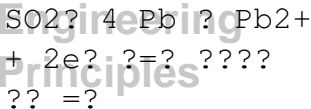
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1/1 Determine
the solubility
product K_{sp} for
 $PbSO_4$. The

desired
equilibrium is
 $PbSO_4 \rightleftharpoons Pb^{2+} +$
 SO_4^{2-} We can
write this as the
sum of two
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equations $PbSO_4$
 $+ 2e^- \rightleftharpoons Pb +$

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