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(pp. 280 – 282) VOCABULARY Complete each sentence.

1. ... Circle the letter of the best answer. 1. Given the balanced equation  $2C_4H_{10} + 13O_2 \rightarrow 8CO_2 + 10H_2O$ , how many moles of  $CO_2$  are produced when 14.9 g of  $O_2$  are used? a. 0.29 mol b. 3.49 mol

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#18, #31, & #38 Answers . 38. To ensure that all  
magnesium is converted to MgO, I would use pure

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oxygen, not air, to carry out the reaction, because Mg could react with  $N_2$  in air to form  $Mg_3N_2$ . The pure oxygen should be in excess. 5. a. 50 mol HI Page 6/24

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The cutting edge of a knife that has been sharpened has a smaller surface area than a dull knife. Since

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pressure is force per unit area, a sharp knife will exert a higher pressure with the same amount of force and cut through material more effectively.

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dioxide and water  $C_5H_{12} + 8O_2 \rightarrow 5CO_2 + 6H_2O$  47.

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Classify the following reactions as single displacement, double displacement, Page 9/26

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